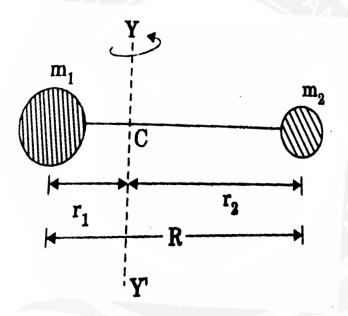
UNIT I

Mechanics

1.4 Moment of Inertia of a diatomic molecule

A diatomic molecule, in its stable equilibrium position consists two atoms that are at a distance 'R' apart. The distance 'R' is called the band length between the two atoms.

Presently we can consider that it consists of two tiny spheres at either end of a thin weightless rigid rod, as shown in fig. This kind of arrangement can be called as rigid rotor.



Let 'C' be the center of mass of the molecule and r_1 and r_2 respective distances of the two atoms from it.

Then

$$r_1 + r_2 = R$$
 ----- (1)

and

$$m_1 r_1 = m_2 r_2$$
 ----- (2)

where m_1 and m_2 are the masses of two atoms respectively.

From eqn. (1),

$$r_1 = R - r_2$$
 ----- (3)

and from eqn. (2),

$$r_2 = \frac{m_1 r_1}{m_2} \qquad ----- (4)$$

so,

$$r_{1} = R - \frac{m_{1}r_{1}}{m_{2}}$$

$$R = r_{1} + \frac{m_{1}r_{1}}{m_{2}} = r_{1} \left[1 + \frac{m_{1}}{m_{2}} \right] - \cdots (5)$$

$$r_{1} = \frac{R}{\left[1 + \frac{m_{1}}{m_{2}} \right]} - \cdots (6)$$

Now, the moment of inertia of the molecule (i.e., of the two atoms) about an axis passing through the centre of mass 'C' and perpendicular to the bond is given as

$$I = m_1 r_1^2 + m_2 r_2^2 ----- (7)$$

$$I = m_1 r_1 r_1 + m_1 r_1 r_2 ---- (6) \quad [:: from eqn. (2)]$$

$$I = m_1 r_1 (r_1 + r_2)$$

(or) by using eqn. (1),

$$I = m_1 r_1 R$$
 ---- (8)

Substituting eqn. (6) in (8) gives

$$I = m_1 R \left[\frac{R}{\left[1 + \frac{m_1}{m_2} \right]} \right]$$

$$I = \frac{m_1 R^2}{\left[1 + \frac{m_1}{m_2} \right]}$$

$$= \frac{m_1 R^2}{\left[\frac{m_2 + m_1}{m_2} \right]}$$

$$= \frac{m_1 m_2 R^2}{m_2 + m_4}$$

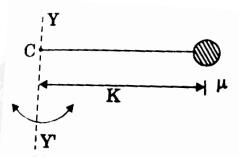
(or)

$$I = \left[\frac{m_1 m_2}{m_2 + m_1}\right] R^2$$

$$I = \mu R^2 \qquad ----- (9)$$

Where $\mu = \frac{m_1 m_2}{m_2 + m_1}$ is called as reduced mass of the molecule. Thus the figure can also be redrawn

as



In figure, K=R, which is called radius of gyration, so moment of inertia

$$I = \mu K^2$$
 ---- (10)

